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Retraction of articles by T. Liu *et al.*

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A series of 29 papers by Liu *et al.* are retracted.

As a result of problems with the data sets and incorrect atom assignments, 29 papers by Liu *et al.* are retracted. Full details of all the articles are given in Table 1.

Table 1
Details of articles to be retracted, in order of publication.

Title	Reference	DOI	Refcode
Tetrakis(pyrazine-κN)bis(thiocyanato-κN)manganese(II)	Liu & Xie (2007*a*)	10.1107/S1600536807026852	EDUMAS
(Dihydroxyglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')copper(II) dinitrate dihydrate	Liu, Wang, Wang & Xie (2007*b*)	10.1107/S1600536807028255	EDUVAB
Tetrakis(pyrazine-κN)bis(thiocyanato-κN)zinc(II)	Liu & Xie (2007*b*)	10.1107/S1600536807028735	RIGQAA
Tetrakis(μ-2-pyridyloxyacetato)bis[(1,10-phenanthroline)(2-pyridyloxyacetato)-lanthanum(III)]	Liu, Wang, Wang & Xie (2007*c*)	10.1107/S1600536807030917	UDUMIQ
Polymeric $KNOF_2$	Liu Wang, Wang & Xie (2007*a*)	10.1107/S1600536807027195	ICSD 240891
(Dihydroxyglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')cobalt(II) dinitrate dihydrate	Liu, Wang, Wang & Xie (2007*d*)	10.1107/S1600536807031224	WIHJED
Tetrakis(μ-2-pyridyloxyacetato)bis[(1,10-phenanthroline)(2-pyridyloxyacetato)-praseodymium(III)]	Liu, Wang, Wang & Xie (2007*e*)	10.1107/S1600536807032679	WIHQEK
Tetrakis[μ-(2-pyridyloxy)acetato-$\kappa^2O:O'$]bis{(1,10-phenanthroline-κ^2N,N')-[(2-pyridyloxy)acetato-κO]neodymium(III)}	Liu, Wang, Wang & Xie (2007*f*)	10.1107/S1600536807035349	TIGDAP
(Dihydroxyglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')manganese(II) dinitrate dihydrate	Liu, Wang, Wang & Xie (2007*g*)	10.1107/S1600536807035076	TIGDET
2-Amino-3,5-dinitrobenzoic acid–ammonia (1/1)	Liu & Zhu (2007*j*)	10.1107/S1600536807040068	KIKQAX
2-Hydroxy-3,5-dinitrobenzamide monohydrate	Liu & Zhu (2007*k*)	10.1107/S1600536807039712	KIKQEB
2-(1-Hydroxy-2-pyridyl)acetamide monohydrate	Liu & Zhu (2007*l*)	10.1107/S1600536807040652	CIKQOD
Bis(2,2′-bipyridine-κN,N′)bis(thiocyanato-κN)iron(II)	Liu & Zhu (2007*a*)	10.1107/S1600536807043486	XIFXOA
catena-Poly[hexakis(μ_2-anilinoacetamide)bis(1,10-phenanthroline)disamarium(III]	Liu & Zhu (2007*b*)	10.1107/S1600536807045485	XILNAI
3-Hydroxy-2,4,6-trinitropyridine monohydrate	Liu & Zhu (2007*m*)	10.1107/S1600536807045230	PILNOO
catena-Poly[hexakis(μ_2-anilinoacetamide)bis(1,10-phenanthroline)-dipraseodymium(III)]	Liu & Zhu (2007*c*)	10.1107/S1600536807047733	SILZET
catena-Poly[[tetra-μ-anilinoacetamidato-bis(1,10-phenanthroline)dicerium(III)]-di-μ-anilinoacetamidato]	Liu & Zhu (2007*d*)	10.1107/S1600536807050969	GIMZOS
Tetrakis(pyridine-κN)bis(thiocyanato-κN)chromium(II)	Liu & Zhu (2007*e*)	10.1107/S1600536807051756	WINFAB
2-Ammonio-3-carboxy-5-nitrobenzoate monohydrate	Liu & Zhu (2007*n*)	10.1107/S1600536807048477	GINFEP
2-(Benzoylhydrazinocarbonyl)benzoic acid	Liu & Zhu (2007*o*)	10.1107/S160053680705204X	TINZIA
Tetrakis(pyridine-κN)bis(thiocyanato-κN)vanadium(II)	Liu & Zhu (2007*f*)	10.1107/S1600536807054529	HIPZIQ
catena-Poly[[(nitrato-κO)(1,10-phenanthroline-κ^2N,N')nickel(II)]-μ-acetamido-$\kappa^2O:N$]	Liu & Zhu (2007*g*)	10.1107/S1600536807056504	XIRGIP
catena-Poly[[(nitrato-κO)(1,10-phenanthroline-κ^2N,N')copper(II)]-μ-acetamido-$\kappa^2O:N$]	Liu & Zhu (2007*h*)	10.1107/S1600536807059077	HIQROP
catena-Poly[[(nitrato-κO)(1,10-phenanthroline-κ^2N,N')cobalt(II)]-μ-acetamidato-$\kappa^2O:N$]	Liu & Zhu (2007*i*)	10.1107/S1600536807060631	YIQMER
N′-Benzoyl-4-nitronicotinohydrazide	Liu & Zhu (2007*p*)	10.1107/S1600536807053068	CIPVON
N′-(3-Nitro-4-pyridylcarbonyl)pyridine-4-carbohydrazide	Liu & Zhu (2007*q*)	10.1107/S1600536807054876	RIRWEV

 doi:10.1107/S1600536809049976

Table 1 (continued)

Title	Reference	DOI	Refcode
Ethylenediammonium sulfate	Liu & Zhu (2007*r*)	10.1107/S1600536807056280	ETDAMS03
Ethylenediammonium perchlorate	Liu & Zhu (2007*s*)	10.1107/S1600536807059909	HIRYEN
catena-Poly[[(nitrato-κO)(1,10-phenanthroline-κ^2N,N')manganese(II)]-μ-nitrato-$\kappa^2O:O'$]	Liu & Zhu (2008)	10.1107/S160053680706254X	MIRROV

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catena-Poly[[(nitrato-κO)(1,10-phenanthroline-$\kappa^2 N,N'$)copper(II)]-μ-acetamido-$\kappa^2 O:N$]

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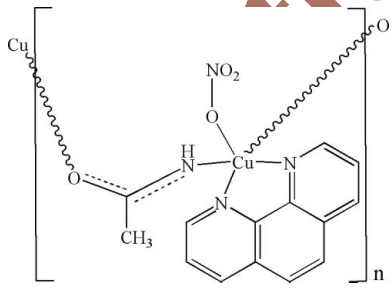
Key indicators: single-crystal X-ray study; T = 273 K; mean σ(C–C) = 0.005 Å; R factor = 0.041; wR factor = 0.130; data-to-parameter ratio = 12.8.

In the crystal structure of the title compound, $[Cu(NO_3)(C_2H_4NO)(C_{12}H_8N_2)]_n$, the Cu^{II} atoms are linked by acetamidate ligands, forming a chain. Each Cu^{II} atom is five-coordinated by two N atoms of the 1,10-phenanthroline ligand, one nitrate O atom, and one N and one O atoms of acetamidate in a trigonal-bipyramidal geometry. In the crystal structure, the chains are linked by hydrogen bonds into a polymeric ribbon structure.

Related literature

For a related structure, see: Liu & Zhu (2007). For bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$[Cu(NO_3)(C_2H_4NO)(C_{12}H_8N_2)]$
M_r = 363.82
Monoclinic, $P2_1/n$
a = 8.680 (2) Å
b = 9.190 (3) Å
c = 17.0137 (12) Å
β = 101.904 (2)°
V = 1328.0 (6) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 1.67 mm^{-1}
T = 273 (2) K
0.41 × 0.23 × 0.22 mm

Data collection

Bruker APEXII area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.624, T_{max} = 0.695
8439 measured reflections
2679 independent reflections
2285 reflections with $I > 2\sigma(I)$
R_{int} = 0.017

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.041
$wR(F^2)$ = 0.130
S = 1.08
2679 reflections
209 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.80 e Å^{-3}
$\Delta\rho_{min}$ = −0.69 e Å^{-3}

Table 1
Selected geometric parameters (Å, °).

Cu1—O1	1.942 (2)	Cu1—N2	1.981 (2)
Cu1—O4^{i}	2.322 (2)	Cu1—N3	2.004 (2)
Cu1—N1	2.011 (3)		
O1—Cu1—O4^{i}	82.50 (9)	O4—Cu1—N2^{i}	125.47 (5)
O1—Cu1—N1	94.00 (11)	O4—Cu1—N3^{i}	141.73 (5)
O1—Cu1—N2	174.61 (10)	N1—Cu1—N2	82.97 (10)
O1—Cu1—N3	91.10 (10)	N1—Cu1—N3	165.98 (10)
O4—Cu1—N1^{i}	138.52 (4)	N2—Cu1—N3	92.87 (10)

Symmetry code: (i) $-x+\frac{3}{2}, y+\frac{1}{2}, -z+\frac{3}{2}$.

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C2—H2···O3^{ii}	0.93	2.50	3.319 (5)	147
C5—H5···O2^{iii}	0.93	2.57	3.171 (5)	124

Symmetry codes: (ii) $-x+\frac{1}{2}, y+\frac{1}{2}, -z+\frac{3}{2}$; (iii) $-x+1, -y+2, -z+2$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: AT2484).

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supplementary materials

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catena-Poly[[(nitrato-κO)(1,10-phenanthroline-$\kappa^2 N,N'$)copper(II)]-μ-acetamido-$\kappa^2 O:N$]

T. Liu and J. Y. Zhu

Comment

The crystal structure of *catena*-Poly[[(nitrato-κ*O*) (1,10-phenanthroline-$\kappa^2 N,N'$)nickel(II)]-μ-acetamido-$\kappa^2 O:N$], (II), has previously been reported (Liu & Zhu, 2007). The crystal structure determination of the title compound, (I), has been carried out in order to elucidate the molecular conformation and to compare it with that of (II). We report herein the crystal structure of (I).

In the molecule of (I) (Fig. 1), the ligand bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The title compound, $[Cu(NO_3)(C_2H_4NO)(C_{12}H_8N_2)]_n$, which are linked by acetamidate ligands to form a chain. Each Cu^{II} atom is five-coordinated by two N atoms of 1,10-phenanthroline (phen) ligand, one nitrate O atom, and one N and one O atoms of acetamidate within a bipyramidal coordination geometry (Table 1). The Cu—O and Cu—N bond are in the range [1.942 (2)–2.322 (2) Å] and [1.981 (2)–2.011 (3) Å], respectively (Table 1), as in (II).

In the crystal structure, no classic C—H···O hydrogen bonds (Fig. 2 and Table 2) seem to be effective in the stabilization of the structure, resulting in the formation of a polymeric ribbon structure, as in (II).

The both compounds, (I) and (II), are isostructural.

Experimental

Crystals of the title compound were synthesized using hydrothermal method in a 23 ml Teflon-lined Parr bomb, which was then sealed. Copper (II) nitrate hexahydrate (296.3 mg, 1 mmol), phen (180.2 mg, 1 mmol), acetamide (59.1 mg, 1 mmol) and distilled water (8 g) were placed into the bomb and sealed. The bomb was then heated under autogenous pressure up to 453 K over the course of 7 d and allowed to cool at room temperature for 24 h. Upon opening the bomb, a clear colorless solution was decanted from small blue crystals. These crystals were washed with distilled water followed by ethanol, and allowed to air-dry at room temperature.

Refinement

The H atoms were positioned geometrically, with N—H = 0.86 Å (for NH), C—H = 0.93 and 0.96 Å for aromatic and methyl H atoms, and constrained to ride on their parent atoms, with $U_{iso}(H) = xU_{eq}(C,N)$, where $x = 1.2$ for aromatic and NH H atoms and $x = 1.5$ for methyl H atoms.

Figures

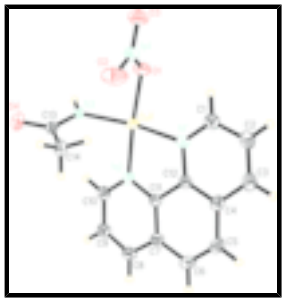

Fig. 1. The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level [symmetry code (A): $-x + 3/2$, $y + 1/2$, $-z + 1/2$].

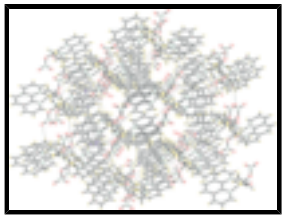

Fig. 2. A packing diagram of (I). Hydrogen bonds are shown as dashed lines.

catena-Poly[[(nitrato-κ*O*)(1,10-phenanthroline- κ^2N,N')copper(II)]-μ-acetamido-$\kappa^2O:N$]

Crystal data

$[Cu(NO_3)(C_2H_4NO)(C_{12}H_8N_2)]$	$F_{000} = 740$
$M_r = 363.82$	$D_x = 1.820$ Mg m^{-3}
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
Hall symbol: -P 2yn	Cell parameters from 5781 reflections
$a = 8.680$ (2) Å	$\theta = 2.3$–27.0º
$b = 9.190$ (3) Å	$\mu = 1.67$ mm^{-1}
$c = 17.0137$ (12) Å	$T = 273$ (2) K
$\beta = 101.904$ (2)º	Prism, blue
$V = 1328.0$ (6) Å^3	0.41 × 0.23 × 0.22 mm
$Z = 4$	

Data collection

Bruker APEXII area-detector diffractometer	2679 independent reflections
Radiation source: fine-focus sealed tube	2285 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{int} = 0.017$
$T = 273(2)$ K	$\theta_{max} = 26.7$º
φ and ω scans	$\theta_{min} = 2.5$º
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$h = -10 \rightarrow 10$
$T_{min} = 0.624$, $T_{max} = 0.695$	$k = -11 \rightarrow 11$
8439 measured reflections	$l = -21 \rightarrow 21$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.041$	H-atom parameters constrained
$wR(F^2) = 0.130$	$w = 1/[\sigma^2(F_o^2) + (0.0915P)^2 + 0.5483P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.08$	$(\Delta/\sigma)_{max} = 0.001$
2679 reflections	$\Delta\rho_{max} = 0.80$ e Å^{-3}
209 parameters	$\Delta\rho_{min} = -0.69$ e Å^{-3}
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Cu1	0.66503 (4)	0.92788 (4)	0.803179 (19)	0.03409 (17)
O1	0.5481 (3)	1.0048 (3)	0.70204 (13)	0.0470 (6)
O2	0.3259 (4)	0.9011 (4)	0.72922 (19)	0.0769 (10)
O3	0.3290 (3)	1.0282 (4)	0.62163 (17)	0.0623 (7)
O4	0.6206 (3)	0.5052 (3)	0.74795 (15)	0.0474 (6)
N1	0.6078 (3)	1.0941 (3)	0.86853 (16)	0.0356 (5)
N2	0.7951 (3)	0.8671 (3)	0.90775 (14)	0.0337 (5)
N3	0.6879 (3)	0.7347 (2)	0.75217 (13)	0.0281 (5)
H3A	0.7419	0.7239	0.7155	0.034*
N4	0.4025 (3)	0.9775 (3)	0.68377 (16)	0.0403 (6)
C1	0.5201 (4)	1.2102 (4)	0.8464 (2)	0.0441 (7)
H1	0.4813	1.2259	0.7919	0.053*
C2	0.4828 (4)	1.3112 (4)	0.9017 (2)	0.0486 (8)
H2	0.4225	1.3930	0.8840	0.058*
C3	0.5364 (4)	1.2873 (4)	0.9812 (2)	0.0474 (8)
H3	0.5094	1.3505	1.0189	0.057*
C4	0.6330 (4)	1.1664 (4)	1.00650 (19)	0.0386 (6)
C5	0.6977 (4)	1.1340 (4)	1.08823 (19)	0.0441 (7)

H5	0.6726	1.1923	1.1285	0.053*
C6	0.7960 (4)	1.0188 (4)	1.10835 (19)	0.0443 (7)
H6	0.8381	1.0004	1.1622	0.053*
C7	0.8363 (4)	0.9251 (3)	1.04838 (18)	0.0364 (6)
C8	0.9427 (4)	0.8075 (4)	1.06354 (19)	0.0430 (7)
H8	0.9926	0.7856	1.1160	0.052*
C9	0.9725 (4)	0.7261 (4)	1.0016 (2)	0.0450 (7)
H9	1.0438	0.6494	1.0115	0.054*
C10	0.8963 (4)	0.7578 (3)	0.92368 (19)	0.0410 (7)
H10	0.9166	0.7012	0.8816	0.049*
C11	0.7675 (3)	0.9512 (3)	0.96854 (17)	0.0313 (6)
C12	0.6653 (3)	1.0726 (3)	0.94749 (18)	0.0319 (6)
C13	0.6164 (3)	0.6283 (3)	0.77790 (16)	0.0298 (5)
C14	0.5453 (4)	0.6522 (3)	0.83151 (19)	0.0384 (7)
H14A	0.4521	0.5932	0.8235	0.058*
H14B	0.5163	0.7531	0.8306	0.058*
H14C	0.6115	0.6292	0.8825	0.058*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cu1	0.0402 (3)	0.0328 (2)	0.0277 (2)	0.00245 (13)	0.00354 (16)	0.00106 (13)
O1	0.0436 (12)	0.0556 (15)	0.0381 (11)	−0.0051 (11)	−0.0004 (9)	0.0129 (11)
O2	0.0571 (16)	0.119 (3)	0.0546 (17)	−0.0216 (17)	0.0123 (14)	0.0260 (18)
O3	0.0540 (14)	0.0697 (17)	0.0541 (16)	−0.0053 (13)	−0.0104 (12)	0.0154 (14)
O4	0.0456 (12)	0.0372 (12)	0.0596 (14)	−0.0041 (10)	0.0115 (11)	−0.0154 (11)
N1	0.0387 (12)	0.0327 (12)	0.0354 (13)	0.0003 (10)	0.0078 (10)	0.0046 (10)
N2	0.0380 (12)	0.0302 (12)	0.0322 (11)	0.0023 (10)	0.0057 (10)	−0.0009 (10)
N3	0.0357 (11)	0.0277 (11)	0.0226 (10)	−0.0040 (9)	0.0097 (9)	−0.0034 (8)
N4	0.0415 (13)	0.0381 (13)	0.0384 (13)	0.0003 (11)	0.0018 (11)	0.0009 (12)
C1	0.0443 (16)	0.0401 (17)	0.0459 (17)	0.0061 (14)	0.0048 (14)	0.0030 (14)
C2	0.0476 (17)	0.0393 (17)	0.059 (2)	0.0113 (14)	0.0112 (15)	0.0014 (15)
C3	0.0483 (17)	0.0381 (17)	0.059 (2)	0.0077 (14)	0.0185 (15)	−0.0056 (15)
C4	0.0392 (14)	0.0383 (15)	0.0408 (15)	−0.0039 (12)	0.0141 (13)	−0.0049 (13)
C5	0.0510 (17)	0.0488 (18)	0.0362 (15)	−0.0016 (15)	0.0177 (14)	−0.0078 (14)
C6	0.0494 (17)	0.0536 (19)	0.0303 (14)	−0.0008 (15)	0.0096 (13)	−0.0018 (14)
C7	0.0408 (15)	0.0374 (16)	0.0311 (14)	−0.0032 (12)	0.0079 (12)	0.0029 (11)
C8	0.0479 (16)	0.0412 (16)	0.0365 (15)	0.0008 (14)	0.0009 (13)	0.0059 (13)
C9	0.0485 (17)	0.0356 (16)	0.0464 (17)	0.0089 (13)	−0.0007 (14)	0.0036 (14)
C10	0.0475 (16)	0.0332 (15)	0.0406 (16)	0.0062 (13)	0.0050 (13)	−0.0037 (12)
C11	0.0348 (13)	0.0288 (13)	0.0301 (13)	−0.0028 (11)	0.0061 (11)	0.0005 (10)
C12	0.0325 (13)	0.0292 (14)	0.0344 (14)	−0.0030 (10)	0.0081 (12)	0.0011 (11)
C13	0.0297 (12)	0.0270 (12)	0.0308 (13)	0.0024 (10)	0.0019 (11)	−0.0028 (10)
C14	0.0509 (16)	0.0299 (14)	0.0446 (15)	−0.0041 (12)	0.0337 (14)	−0.0055 (12)

Geometric parameters (Å, °)

Cu1—O1	1.942 (2)	C3—C4	1.405 (5)

Cu1—O4^{i}	2.322 (2)	C3—H3	0.9300
Cu1—N1	2.011 (3)	C4—C12	1.395 (4)
Cu1—N2	1.981 (2)	C4—C5	1.420 (5)
Cu1—N3	2.004 (2)	C5—C6	1.359 (5)
O1—N4	1.263 (4)	C5—H5	0.9300
O2—N4	1.321 (4)	C6—C7	1.432 (4)
O3—N4	1.210 (4)	C6—H6	0.9300
O4—C13	1.244 (4)	C7—C11	1.387 (4)
O4—Cu1^{ii}	2.322 (2)	C7—C8	1.411 (5)
N1—C1	1.320 (4)	C8—C9	1.360 (5)
N1—C12	1.347 (4)	C8—H8	0.9300
N2—C10	1.325 (4)	C9—C10	1.385 (5)
N2—C11	1.352 (4)	C9—H9	0.9300
N3—C13	1.283 (4)	C10—H10	0.9300
N3—H3A	0.8600	C11—C12	1.425 (4)
C1—C2	1.406 (5)	C13—C14	1.222 (4)
C1—H1	0.9300	C14—H14A	0.9600
C2—C3	1.355 (5)	C14—H14B	0.9600
C2—H2	0.9300	C14—H14C	0.9600
O1—Cu1—O4^{i}	82.50 (9)	C3—C4—C5	123.8 (3)
O1—Cu1—N1	94.00 (11)	C6—C5—C4	120.8 (3)
O1—Cu1—N2	174.61 (10)	C6—C5—H5	119.6
O1—Cu1—N3	91.10 (10)	C4—C5—H5	119.6
O4—Cu1—N1^{i}	138.52 (4)	C5—C6—C7	121.4 (3)
O4—Cu1—N2^{i}	125.47 (5)	C5—C6—H6	119.3
O4—Cu1—N3^{i}	141.73 (5)	C7—C6—H6	119.3
N1—Cu1—N2	82.97 (10)	C11—C7—C8	116.4 (3)
N1—Cu1—N3	165.98 (10)	C11—C7—C6	118.4 (3)
N2—Cu1—N3	92.87 (10)	C8—C7—C6	125.2 (3)
N4—O1—Cu1	117.34 (19)	C9—C8—C7	120.1 (3)
C13—O4—Cu1^{ii}	121.63 (19)	C9—C8—H8	119.9
C1—N1—C12	118.4 (3)	C7—C8—H8	119.9
C1—N1—Cu1	130.6 (2)	C8—C9—C10	119.8 (3)
C12—N1—Cu1	110.93 (19)	C8—C9—H9	120.1
C10—N2—C11	119.5 (3)	C10—C9—H9	120.1
C10—N2—Cu1	128.9 (2)	N2—C10—C9	121.4 (3)
C11—N2—Cu1	111.59 (19)	N2—C10—H10	119.3
C13—N3—Cu1	115.46 (18)	C9—C10—H10	119.3
C13—N3—H3A	122.3	N2—C11—C7	122.9 (3)
Cu1—N3—H3A	122.3	N2—C11—C12	117.1 (3)
O3—N4—O2	118.6 (3)	C7—C11—C12	120.1 (3)
O3—N4—O1	117.3 (3)	N1—C12—C4	122.6 (3)
O2—N4—O1	124.1 (3)	N1—C12—C11	116.6 (3)
N1—C1—C2	122.9 (3)	C4—C12—C11	120.7 (3)
N1—C1—H1	118.6	C14—C13—O4	122.1 (3)
C2—C1—H1	118.6	C14—C13—N3	118.1 (3)
C3—C2—C1	118.8 (3)	O4—C13—N3	119.8 (3)

C3—C2—H2	120.6	C13—C14—H14A	109.5
C1—C2—H2	120.6	C13—C14—H14B	109.5
C2—C3—C4	119.6 (3)	H14A—C14—H14B	109.5
C2—C3—H3	120.2	C13—C14—H14C	109.5
C4—C3—H3	120.2	H14A—C14—H14C	109.5
C12—C4—C3	117.7 (3)	H14B—C14—H14C	109.5
C12—C4—C5	118.5 (3)		

Symmetry codes: (i) −*x*+3/2, *y*+1/2, −*z*+3/2; (ii) −*x*+3/2, *y*−1/2, −*z*+3/2.

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C2—H2···O3^{iii}	0.93	2.50	3.319 (5)	147
C5—H5···O2^{iv}	0.93	2.57	3.171 (5)	124

Symmetry codes: (iii) −*x*+1/2, *y*+1/2, −*z*+3/2; (iv) −*x*+1, −*y*+2, −*z*+2.

Fig. 1

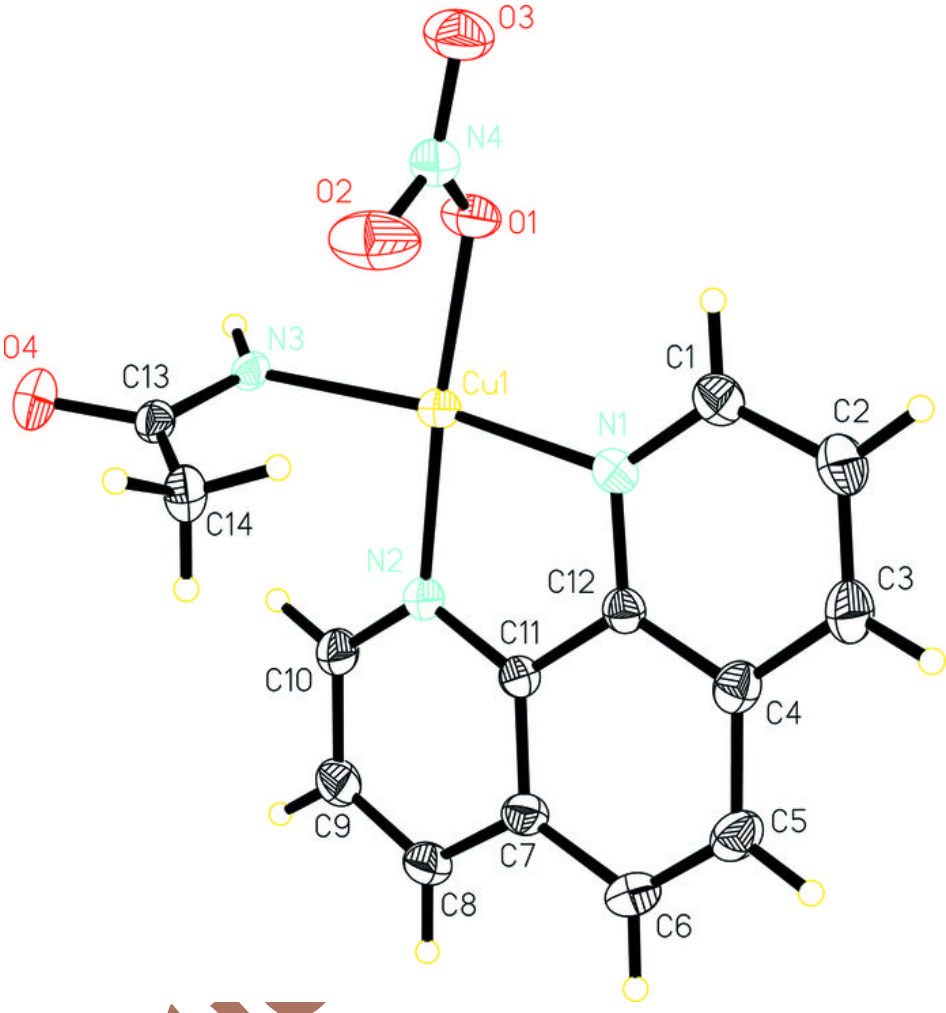
O3
N4
O2
O1
O4
N3
C13
Cu1
C1
N1
C2
C14
N2
C12
C3
C11
C10
C4
C9
C5
C7
C8
C6

Fig. 2